

# The Tri- $\pi$ -Methane Rearrangement; Mechanistic and Exploratory Organic Photochemistry

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## Supporting Information

### I. Derivation of Equation 4

$$1) \quad dA/dt = -(k_1 + k_2)A$$

$$2) \quad A = A_0 e^{-(k_1 + k_2)t}$$

$$3) \quad dB/dt = k_1A - k_3B = k_1A_0 e^{-(k_1 + k_2)t} - k_3B$$

$$4) \quad dB/dt + k_3B = k_1A = k_1A_0 e^{-(k_1 + k_2)t}$$

$$5) \quad \text{take } B = uv \quad \text{then } dB/dt = u(dv/dt) + v(du/dt)$$

$$6) \quad u(dv/dt) + v(du/dt) + k_3uv = k_1A_0 e^{-(k_1 + k_2)t}$$

$$7) \quad u(dv/dt + k_3v) + v(du/dt) = k_1A_0 e^{-(k_1 + k_2)t}$$

$$8) \quad \text{take } dv/dt + k_3v = 0, \text{ then } \int dv/v = -\int k_3 dt$$

$$9) \quad \ln v = \ln v_0 - k_3 t \quad \text{or} \quad v = v_0 e^{-k_3 t}$$

$$10) \quad v_0 e^{-k_3 t} (du/dt) = k_1 A_0 e^{-(k_1 + k_2)t}$$

$$11) \quad \int du = \int (k_1 A_0 / v_0) e^{(k_3 - k_1 - k_2)t} dt$$

$$12) \quad u = u_0 + [k_1 A_0 / v_0 (k_3 - k_1 - k_2)] e^{(k_3 - k_1 - k_2)t}$$

$$13) \quad B = uv = u_0 v_0 e^{-k_3 t} + [k_1 A_0 / (k_3 - k_1 - k_2)] e^{-(k_1 + k_2)t} - [k_1 A_0 / (k_3 - k_1 - k_2)] e^{-k_3 t}$$

$$14) \quad \text{For } B_0 = 0, \text{ then}$$

$$15) \quad B = (k_1 A_0) / (k_3 - k_1 - k_2) [e^{-(k_1 + k_2)t} - e^{-k_3 t}]$$

## II. Photolysis Conditions

Photolyses were carried out with three filter cells of 2.0 M NiSO<sub>4</sub> in 5% H<sub>2</sub>SO<sub>4</sub>, 0.004 M SnCl<sub>2</sub> in 15% HCl and 0.8 M CoSO<sub>4</sub> in 5% H<sub>2</sub>SO<sub>4</sub>. This gave a band pass of 300-0350 nm suitable for selective naphthalene absorption and also for diphenylvinyl absorption in the absence of naphthalene.